Engineering cathode for limiting polysulfide shuttling in **Lithium Sulfur batteries**

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Introduction

- Lithium sulfur (LiS) batteries are highenergy rechargeable batteries that use Li as anode and S as cathode
- They possess high practical energy density up to 500 Wh/kg outperforming Li-ion batteries, 200 – 300 Wh/kg¹

Challenge: Liquid phase conversion of sulfides allows dissolution of long-chain polysulfides (LiPSs) (shuttling effect). However, solid phase conversion prevents formation of long chain LiPSs

Aim: Immobilizing short chain polysulfides in pores of activated carbon (C) host to limit shuttling effect

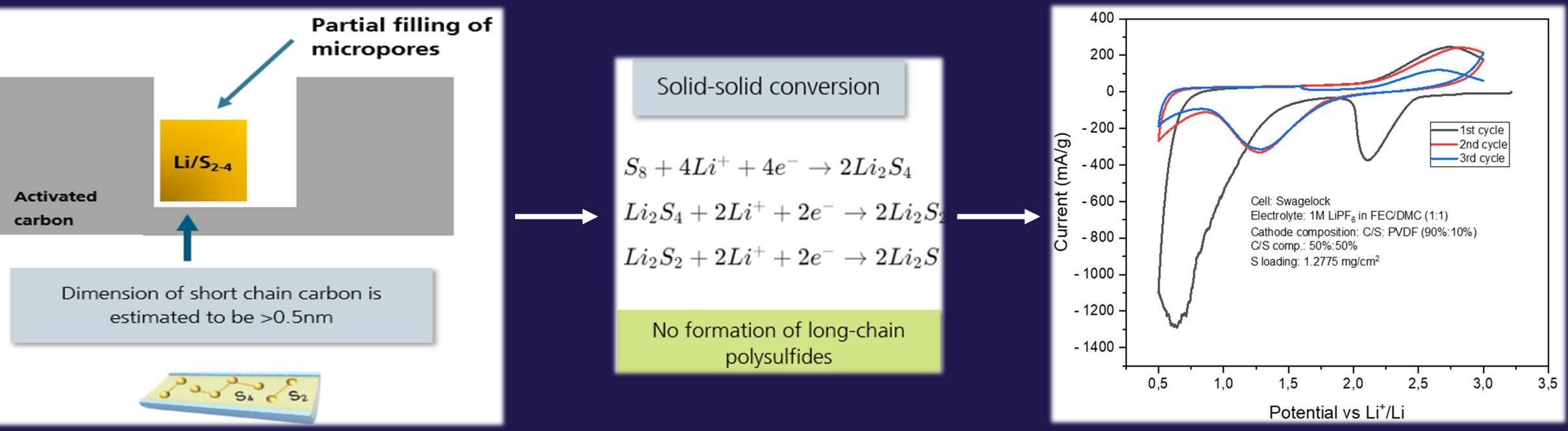
Strategy: Carbon pore engineering and cathode electrolyte interface formation enabling solid phase conversion

Methods

- Screening of carbon samples using BET and pore size distribution
- Melt infiltration of C & S (155°C, 21h)
- Positive electrode was prepared as a 3. slurry mixture composition C/S:PVDF (90:10), using DMSO as a solvent
- Assembling 3-electrode Swagelok cell to 4. evaluate electrochemical performance of C/S cathode

Results

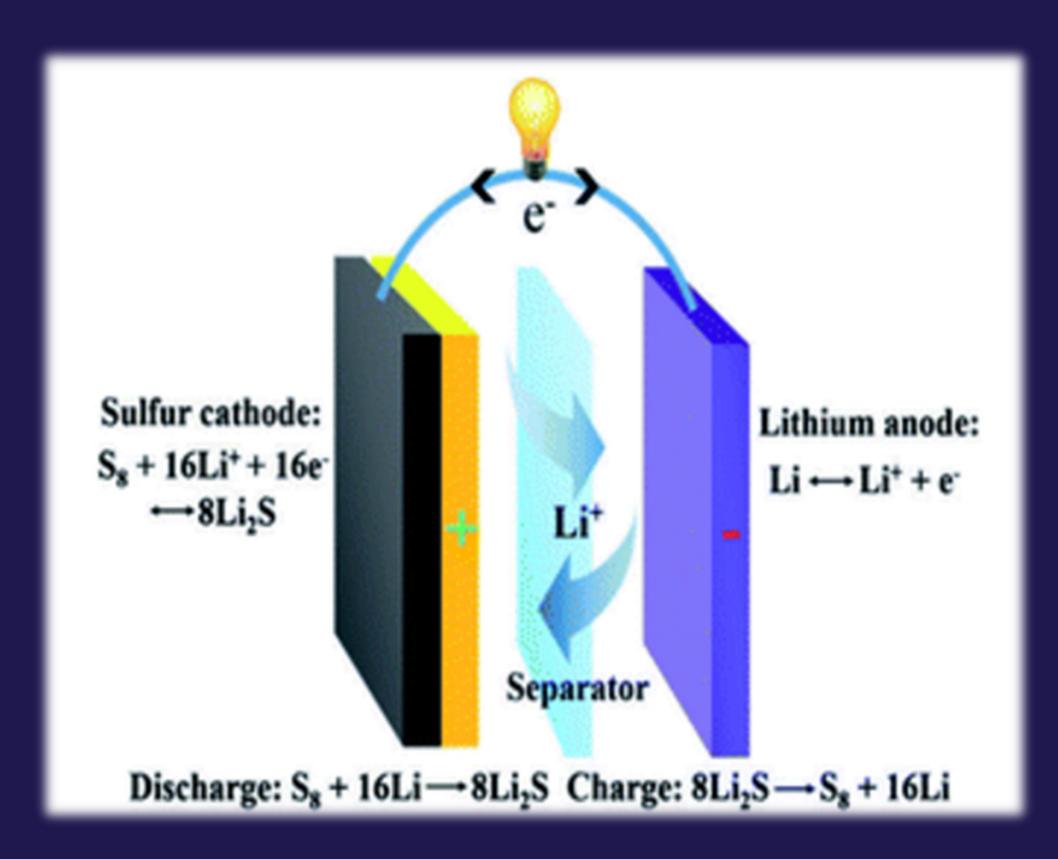
- Solid phase conversion achieved due to single redox peaks.
- High polarization between oxidation and reduction reactions, capacity loss due to parasitic reactions at the slow current rate (C/40)
- High resistance due to CEI formation after the first cycle
- 1 Zhao et al., ACS Cent. Sci. 2020, 6, 7, 1095–1104
- 2 Kim et al., J. Mater. Chem. A, 2019,7, 2942-2964







Carbon host with micropores of 0.8nm hosts short chain polysulfides, enabling solid phase conversion



Schematic of Lithium sulfur battery with redox reaction²



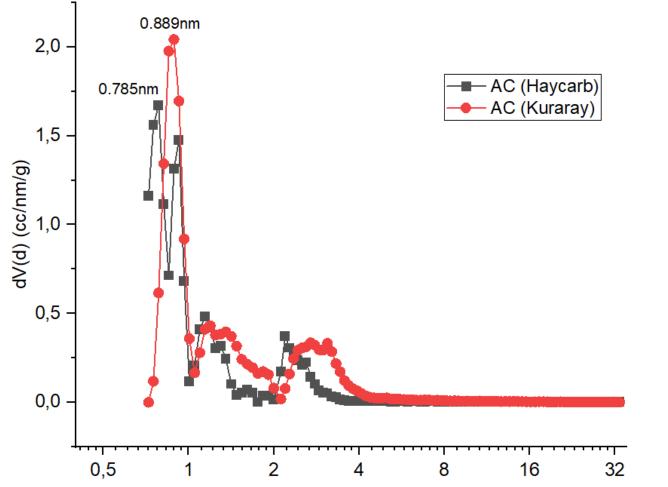
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Cyclic voltammetry (CV) of C/S cathode

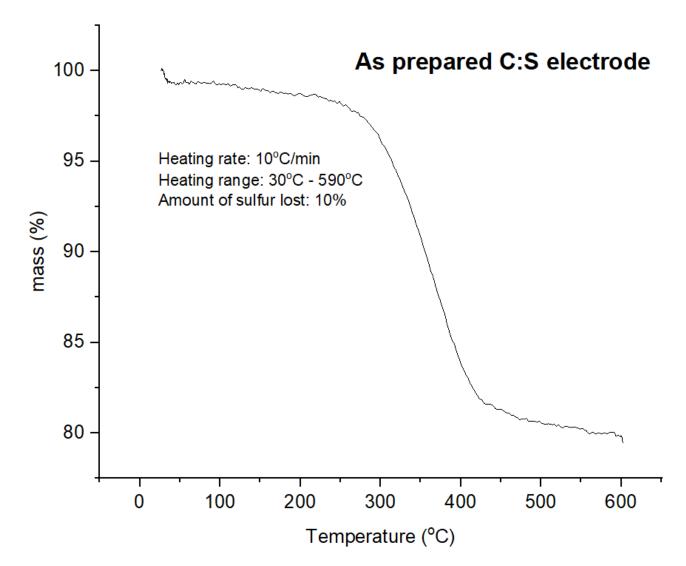
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Characterization

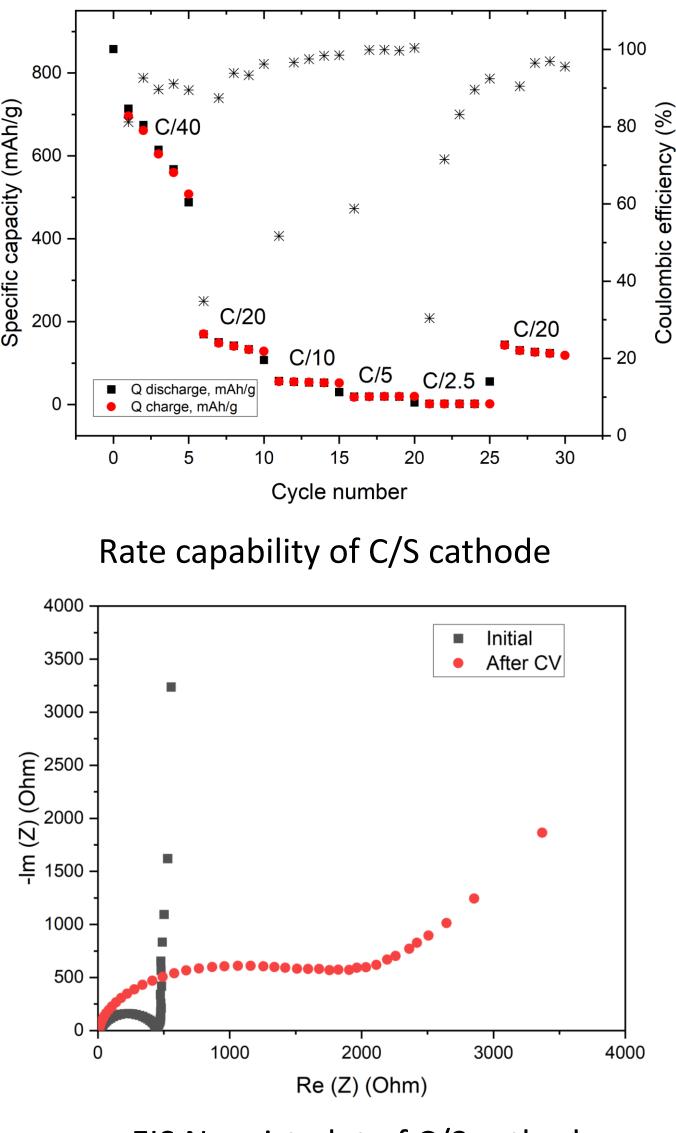


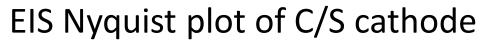
Pore size distribution of carbon samples via BET analysis



Thermogravimetric analysis of as prepared C/S electrode

Cathode Performance





Acknowledgement

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