Investigations of Sulfur-Containing Organics/Polymers as Cathodes and Additives in Rechargeable Sodium–Sulfur Batteries

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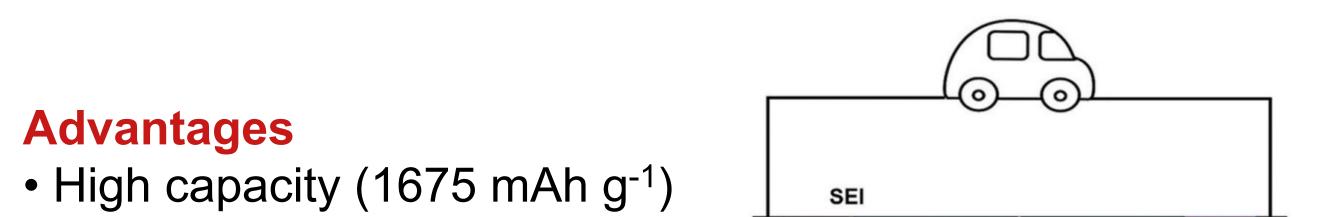
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Sodium-Sulfur Battery (Na-S Battery)

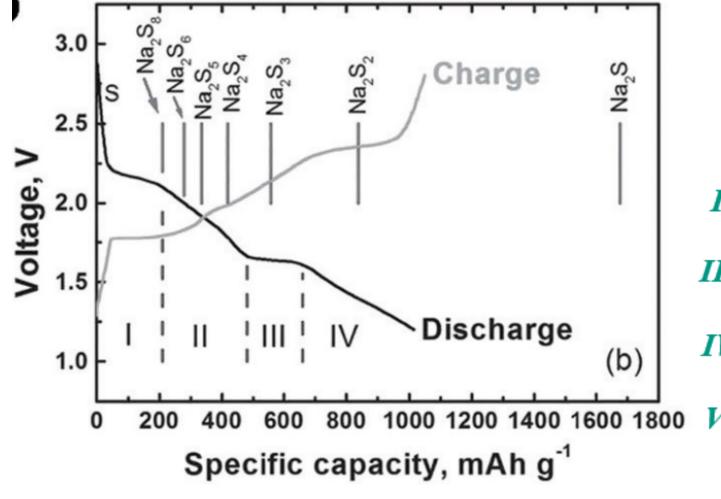


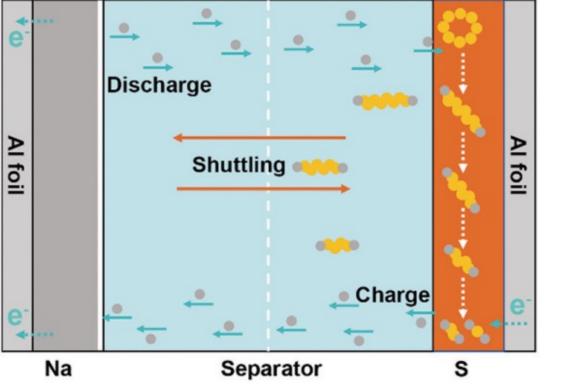
Challenges of Na-S Battery

- Electronic insulation: Hinder charge conduction
- Shuttle effect: loss of sulfur, capacity attenuation, degradation of sodium anode
- Volume change effect: electrode degradation

- Rich reserves for sodium and sulfur
- Low cost
- Environment-friendlyness

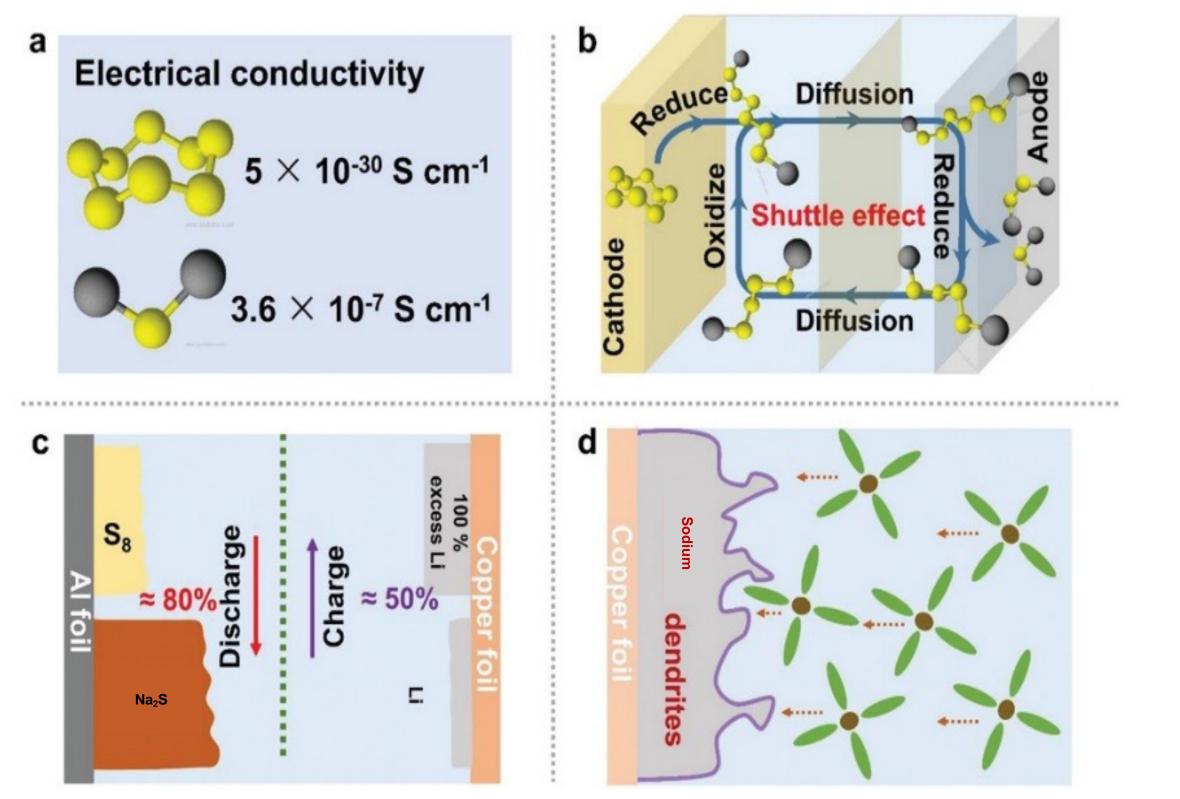
Working mechanism





- solid-liquid-solid
- $S_8+2Na^++2e^-\rightarrow Na_2S_8$ (2.2V)
- $II Na_2S_8 + 2Na^+ + 2e^- \rightarrow 2Na_2S_4 (2.2~1.65 V)$ $\mathbf{III} 3Na_2S_4 + 2Na^+ + 2e^- \rightarrow 4Na_2S_3 \quad (1.65 \text{ V})$ IV Na₂S₄+2Na⁺+2e⁻ \rightarrow 2Na₂S₂ (1.65 V)
- $Na_2S_2+2Na^++2e^-\rightarrow 2Na_2S$ (1.65~1.2 V)

• Sodium dendrites: short circuit, potential safety hazard



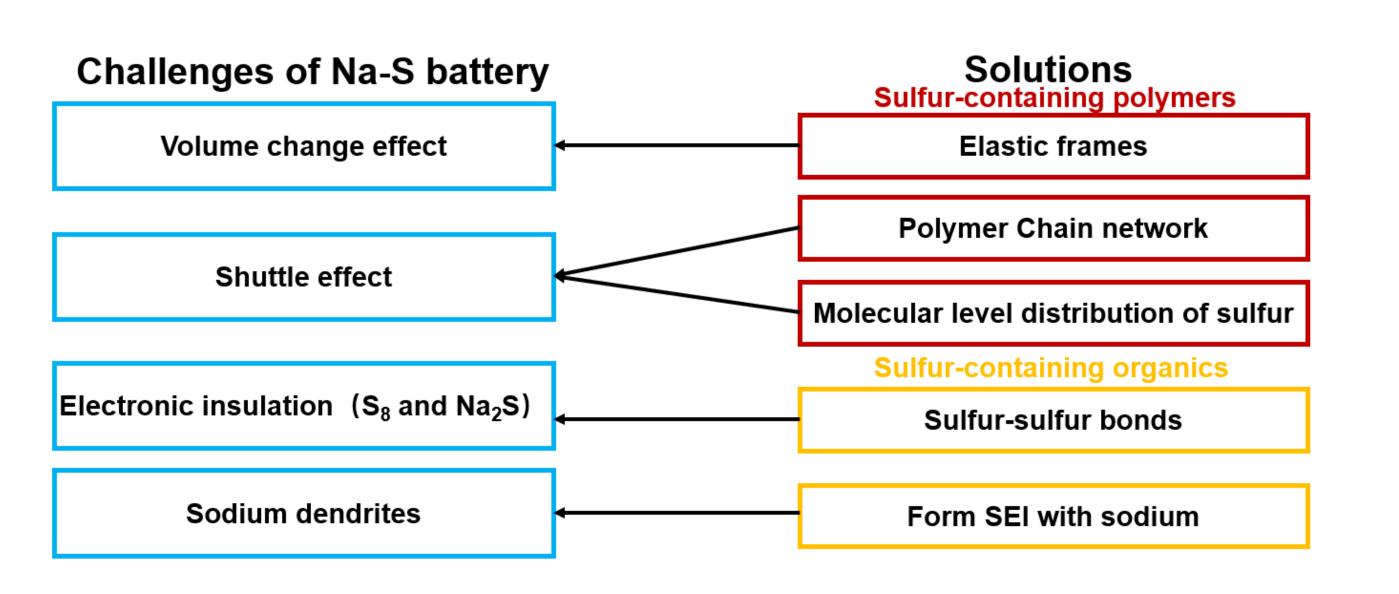
Energy & Environmental Science, 2015, 8 (12): 3477-3494

Solve Challenges by Sulfur-Containing Polymer/Organics

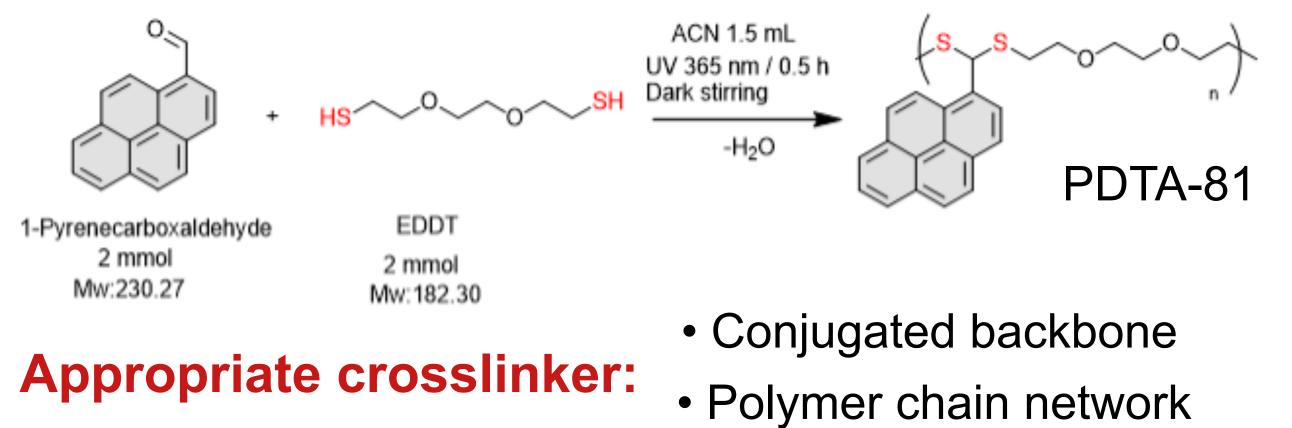
Characterization

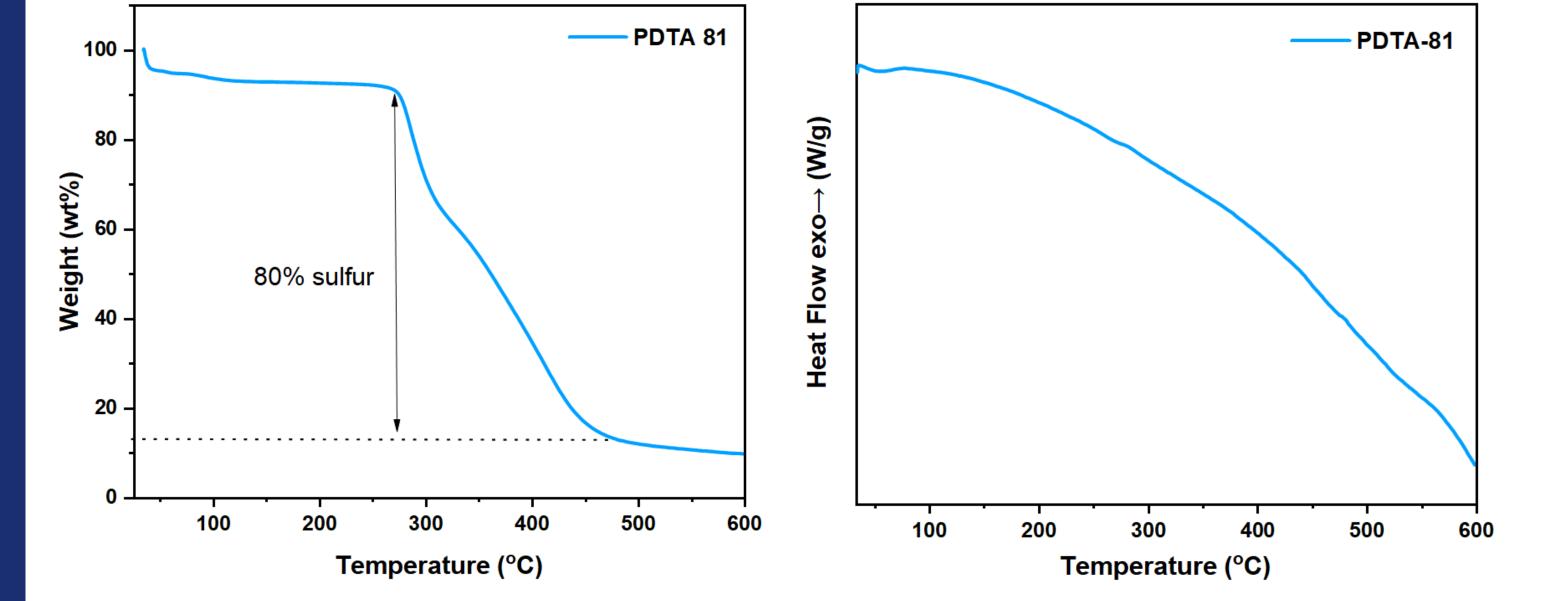
• TGA

• DSC



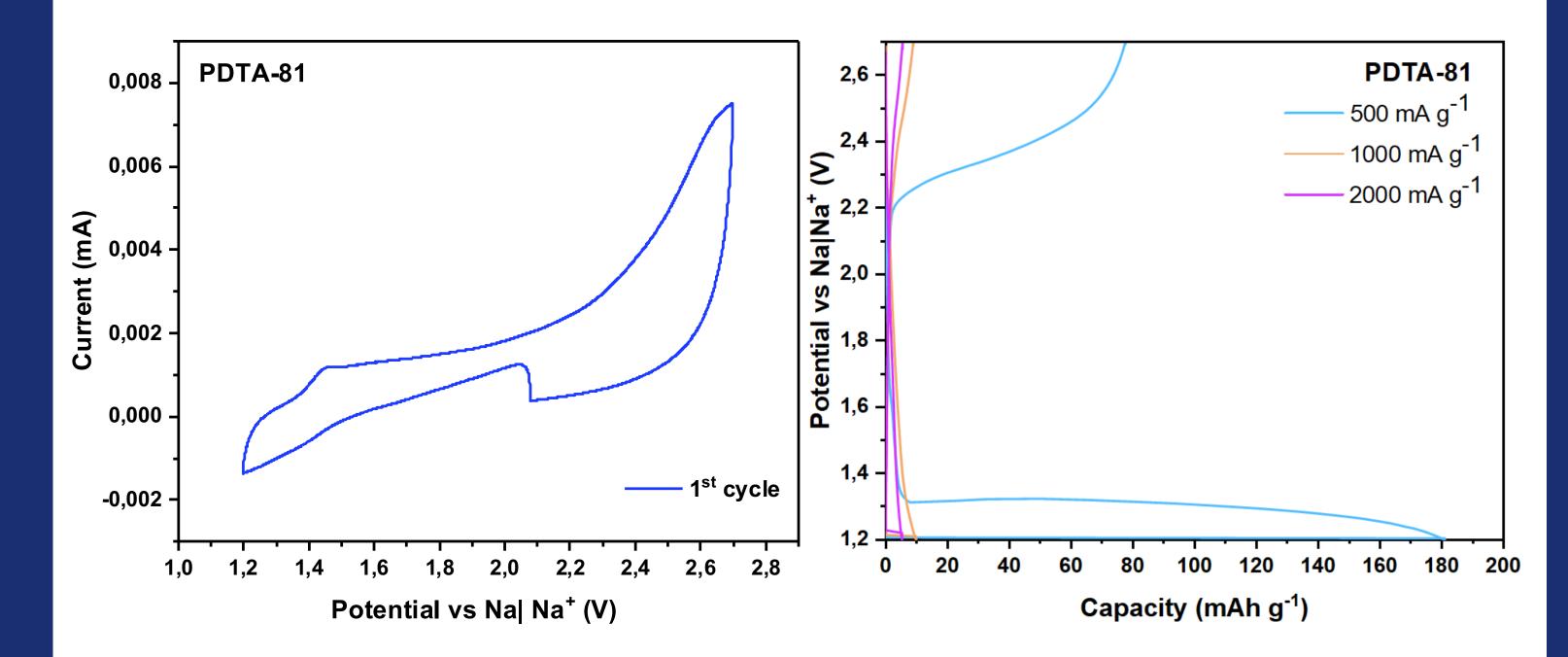
Material Synthesis





Electrochemical performance test

Working electrode: PDTA-81: C65:LA133=6:3:1 (wt) **Separator:** glass fiber **Electrolyte:**NaCF₃SO₃ in TEGDME **Counter electrode:** Sodium **Reference electrode:** Sodium



Download the poster

Plan and Outlook

- Improve the capacity of Na-S battery with polymer cathode.
- Explore the working mechanism of polymer as cathode material.

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